

Number	Reaction	<i>A</i>	<i>n</i>	<i>E</i>	Ref.
1f	$H + O_2 \rightleftharpoons OH + O$	3.520E+16	-0.70	71.4	[1]
2f	$H_2 + O \rightleftharpoons OH + H$	5.060E+04	2.67	26.3	[1]
3f	$H_2 + OH \rightleftharpoons H_2O + H$	1.170E+09	1.30	15.2	[1]
4f	$H_2O + O \rightleftharpoons 2 OH$	7.600E+00	3.84	53.5	[1]
a5f ^a	$H + O + M^{(2)} \rightleftharpoons OH + M^{(2)}$	6.200E+16	-0.60	0	[2, 3]
a6f	$H_2 + O_2 \rightleftharpoons 2 OH$	1.700E+13	0.00	200	[4, 3]
5f ^a	$2 H + M^{(1)} \rightleftharpoons H_2 + M^{(1)}$	7.200E+17	-1.00	0	[1]
6f ^a	$H + OH + M^{(2)} \rightleftharpoons H_2O + M^{(2)}$	2.200E+22	-2.00	0	[1]
7f ^a	$2 O + M^{(2)} \rightleftharpoons O_2 + M^{(2)}$	6.170E+15	-0.50	0	[1]
8f ^{a,b}	$H + O_2 + M^{(6)} \rightleftharpoons HO_2 + M^{(6)}$	k_0 2.600E+19 k_∞ 4.650E+12	-1.20 0.44	0	[5, 6]
a11f ^a	$O + OH + M \rightleftharpoons HO_2 + M$	1.000E+16	0.00	0	[4, 3]
9f	$HO_2 + H \rightleftharpoons 2 OH$	1.700E+14	0.00	3.66	[1]
10f	$HO_2 + H \rightleftharpoons H_2 + O_2$	4.280E+13	0.00	5.9	[1]
11f	$HO_2 + H \rightleftharpoons H_2O + O$	3.100E+13	0.00	7.2	[1]
12f	$HO_2 + O \rightleftharpoons OH + O_2$	2.000E+13	0.00	0	[1]
13f	$HO_2 + OH \rightleftharpoons H_2O + O_2$	2.890E+13	0.00	-2.08	[1]
14f ^{a,b}	$2 OH + M^{(7)} \rightleftharpoons H_2O_2 + M^{(7)}$	k_0 2.300E+18 k_∞ 7.400E+13	-0.90 -0.37	-7.12 0	[5]
15f	$2 HO_2 \rightleftharpoons H_2O_2 + O_2$	3.020E+12	0.00	5.8	[1]
16f	$H_2O_2 + H \rightleftharpoons HO_2 + H_2$	4.790E+13	0.00	33.3	[1]
17f	$H_2O_2 + H \rightleftharpoons H_2O + OH$	1.000E+13	0.00	15	[1]
18f	$H_2O_2 + OH \rightleftharpoons H_2O + HO_2$	7.080E+12	0.00	6	[1]
19f	$H_2O_2 + O \rightleftharpoons HO_2 + OH$	9.630E+06	2.00	16.7	[1]
20f	$CO + OH \rightleftharpoons CO_2 + H$	4.400E+06	1.50	-3.1	[1]
21f	$CO + HO_2 \rightleftharpoons CO_2 + OH$	6.030E+13	0.00	96	[1]
22f ^a	$CHO + M^{(4)} \rightleftharpoons CO + H + M^{(4)}$	1.860E+17	-1.00	71.1	[7]
23f	$CHO + H \rightleftharpoons CO + H_2$	1.000E+14	0.00	0	[1]
24f	$CHO + O \rightleftharpoons CO + OH$	3.000E+13	0.00	0	[1]
25f	$CHO + O \rightleftharpoons CO_2 + H$	3.000E+13	0.00	0	[1]
26f	$CHO + OH \rightleftharpoons CO + H_2O$	5.020E+13	0.00	0	[1]
27f	$CHO + O_2 \rightleftharpoons CO + HO_2$	3.000E+12	0.00	0	[1]
28f ^a	$CH_2O + M^{(1)} \rightleftharpoons CHO + H + M^{(1)}$	6.260E+16	0.00	326	[1]
29f	$CH_2O + H \rightleftharpoons CHO + H_2$	1.260E+08	1.62	9.06	[1]
30f	$CH_2O + O \rightleftharpoons CHO + OH$	3.500E+13	0.00	14.7	[1]
31f	$CH_2O + OH \rightleftharpoons CHO + H_2O$	3.900E+10	0.89	1.7	[1]

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32f	$\text{CH}_4 + \text{H} \rightleftharpoons \text{H}_2 + \text{CH}_3$	1.300E+04	3.00	33.6	[8]	
33f	$\text{CH}_4 + \text{OH} \rightleftharpoons \text{H}_2\text{O} + \text{CH}_3$	1.600E+07	1.83	11.6	[8]	
34f	$\text{CH}_4 + \text{O} \rightleftharpoons \text{CH}_3 + \text{OH}$	1.900E+09	1.44	36.3	[9]	
35f	$\text{CH}_4 + \text{O}_2 \rightleftharpoons \text{CH}_3 + \text{HO}_2$	3.980E+13	0.00	238	[7, 10]	
36f	$\text{CH}_4 + \text{HO}_2 \rightleftharpoons \text{CH}_3 + \text{H}_2\text{O}_2$	9.030E+12	0.00	103	[7, 10]	
37f	$\text{CH}_3 + \text{H} \rightleftharpoons \text{T-CH}_2 + \text{H}_2$	1.800E+14	0.00	63.2	[9]	
38f	$\text{CH}_3 + \text{H} \rightleftharpoons \text{S-CH}_2 + \text{H}_2$	1.550E+14	0.00	56.4	[9]	
39f	$\text{CH}_3 + \text{OH} \rightleftharpoons \text{S-CH}_2 + \text{H}_2\text{O}$	1.000E+13	0.00	10.5	[11]	
40f	$\text{CH}_3 + \text{O} \rightleftharpoons \text{CH}_2\text{O} + \text{H}$	8.430E+13	0.00	0	[9]	
41f	$\text{CH}_3 + \text{T-CH}_2 \rightleftharpoons \text{C}_2\text{H}_4 + \text{H}$	4.220E+13	0.00	0	[12]	
42f	$\text{CH}_3 + \text{HO}_2 \rightleftharpoons \text{CH}_3\text{O} + \text{OH}$	2.000E+13	0.00	0	[9]	
43f	$\text{CH}_3 + \text{O}_2 \rightleftharpoons \text{CH}_2\text{O} + \text{OH}$	3.300E+11	0.00	37.4	[12]	
44f	$\text{CH}_3 + \text{O}_2 \rightleftharpoons \text{CH}_3\text{O} + \text{O}$	1.330E+14	0.00	131	[12]	
45f	$2 \text{CH}_3 \rightleftharpoons \text{C}_2\text{H}_4 + \text{H}_2$	1.000E+14	0.00	134	[13]	
46f	$2 \text{CH}_3 \rightleftharpoons \text{C}_2\text{H}_5 + \text{H}$	3.160E+13	0.00	61.5	[14]	
47f	$\text{CH}_3 + \text{H} \rightleftharpoons \text{CH}_4$	k_0 k_∞	6.260E+23 2.110E+14	-1.80 0.00	0 0	[15]
48f	$2 \text{CH}_3 \rightleftharpoons \text{C}_2\text{H}_6$	k_0 k_∞	1.270E+41 1.810E+13	-7.00 0.00	11.6 0	[8]
m1f	$\text{CH}_3\text{OH} + \text{OH} \rightleftharpoons \text{CH}_2\text{OH} + \text{H}_2\text{O}$	1.440E+06	2.00	-3.51	[16]	
m2f	$\text{CH}_3\text{OH} + \text{OH} \rightleftharpoons \text{CH}_3\text{O} + \text{H}_2\text{O}$	6.300E+06	2.00	6.3	[16]	
m3f	$\text{CH}_3\text{OH} + \text{H} \rightleftharpoons \text{CH}_2\text{OH} + \text{H}_2$	1.640E+07	2.00	18.9	[16]	
m4f	$\text{CH}_3\text{OH} + \text{H} \rightleftharpoons \text{CH}_3\text{O} + \text{H}_2$	3.830E+07	2.00	24.5	[16]	
m5f	$\text{CH}_3\text{OH} + \text{O} \rightleftharpoons \text{CH}_2\text{OH} + \text{OH}$	1.000E+13	0.00	19.6	[16]	
m6f	$\text{CH}_3\text{OH} + \text{HO}_2 \rightleftharpoons \text{CH}_2\text{OH} + \text{H}_2\text{O}_2$	6.200E+12	0.00	81.1	[16]	
m7f	$\text{CH}_3\text{OH} + \text{O}_2 \rightleftharpoons \text{CH}_2\text{OH} + \text{HO}_2$	2.000E+13	0.00	188	[16]	
49f	$\text{S-CH}_2 + \text{OH} \rightleftharpoons \text{CH}_2\text{O} + \text{H}$	3.000E+13	0.00	0	[9]	
50f	$\text{S-CH}_2 + \text{O}_2 \rightleftharpoons \text{CO} + \text{OH} + \text{H}$	3.130E+13	0.00	0	[9]	
51f	$\text{S-CH}_2 + \text{CO}_2 \rightleftharpoons \text{CO} + \text{CH}_2\text{O}$	3.000E+12	0.00	0	[17]	
52f ^a	$\text{S-CH}_2 + \text{M}^{(5)} \rightleftharpoons \text{T-CH}_2 + \text{M}^{(5)}$	6.000E+12	0.00	0	[9]	
53f	$\text{T-CH}_2 + \text{H} \rightleftharpoons \text{CH} + \text{H}_2$	6.020E+12	0.00	-7.48	[12]	
54f	$\text{T-CH}_2 + \text{OH} \rightleftharpoons \text{CH}_2\text{O} + \text{H}$	2.500E+13	0.00	0	[9]	
55f	$\text{T-CH}_2 + \text{OH} \rightleftharpoons \text{CH} + \text{H}_2\text{O}$	1.130E+07	2.00	12.6	[9]	
56f	$\text{T-CH}_2 + \text{O} \rightleftharpoons \text{CO} + 2 \text{H}$	8.000E+13	0.00	0	[18]	
57f	$\text{T-CH}_2 + \text{O} \rightleftharpoons \text{CO} + \text{H}_2$	4.000E+13	0.00	0	[18]	
58f	$\text{T-CH}_2 + \text{O}_2 \rightleftharpoons \text{CO}_2 + \text{H}_2$	2.630E+13	0.00	6.24	[17]	

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59f	T-CH ₂ + O ₂ ⇌ CO + OH + H	6.580E+13	0.00	6.24	[17]	
60f	2 T-CH ₂ ⇌ C ₂ H ₂ + 2 H	1.000E+14	0.00	0	[9]	
61f	CH + O ⇌ CO + H	4.000E+13	0.00	0	[19]	
62f	CH + O ₂ ⇌ CHO + O	1.770E+11	0.76	-2	[20]	
63f	CH + H ₂ O ⇌ CH ₂ O + H	1.170E+15	-0.75	0	[17]	
64f	CH + CO ₂ ⇌ CHO + CO	4.800E+01	3.22	-13.5	[20]	
65f	CH ₂ OH + H ⇌ CH ₂ O + H ₂	3.000E+13	0.00	0	[16]	
66f	CH ₂ OH + H ⇌ CH ₃ + OH	1.750E+14	0.00	11.7	[16]	
67f	CH ₂ OH + OH ⇌ CH ₂ O + H ₂ O	2.400E+13	0.00	0	[16]	
68f	CH ₂ OH + O ₂ ⇌ CH ₂ O + HO ₂	5.000E+12	0.00	0	[16]	
69f ^a	CH ₂ OH + M ⁽⁵⁾ ⇌ CH ₂ O + H + M ⁽⁵⁾	5.000E+13	0.00	105	[16]	
70f	CH ₃ O + H ⇌ CH ₂ O + H ₂	2.000E+13	0.00	0	[16]	
71f	CH ₃ O + H ⇌ S-CH ₂ + H ₂ O	1.600E+13	0.00	0	[16]	
72f	CH ₃ O + OH ⇌ CH ₂ O + H ₂ O	5.000E+12	0.00	0	[16]	
73f	CH ₃ O + O ⇌ OH + CH ₂ O	1.000E+13	0.00	0	[16]	
74f	CH ₃ O + O ₂ → CH ₂ O + HO ₂	4.280E-13	7.60	-14.8	[16]	
75f ^a	CH ₃ O + M ⇌ CH ₂ O + H + M	1.000E+13	0.00	56.5	[16]	
76f ^a	CH ₃ O + M ⁽²⁾ ⇌ CH ₂ OH + M ⁽²⁾	1.000E+14	0.00	80	[16]	
77f	C ₂ H ₆ + H ⇌ C ₂ H ₅ + H ₂	5.400E+02	3.50	21.8	[9]	
78f	C ₂ H ₆ + O ⇌ C ₂ H ₅ + OH	1.400E+00	4.30	11.6	[9]	
79f	C ₂ H ₆ + OH ⇌ C ₂ H ₅ + H ₂ O	2.200E+07	1.90	4.7	[9]	
80f	C ₂ H ₆ + CH ₃ ⇌ C ₂ H ₅ + CH ₄	5.500E-01	4.00	34.7	[9]	
81f	C ₂ H ₆ ⇌ C ₂ H ₅ + H	<i>k</i> ₀	4.900E+42	-6.43	448	[8]
		<i>k</i> _∞	8.850E+20	-1.23	428	
82f	C ₂ H ₅ + H ⇌ C ₂ H ₄ + H ₂	3.000E+13	0.00	0	[9]	
83f	C ₂ H ₅ + O ⇌ C ₂ H ₄ + OH	3.060E+13	0.00	0	[9]	
84f	C ₂ H ₅ + O ⇌ CH ₃ + CH ₂ O	4.240E+13	0.00	0	[9]	
85f	C ₂ H ₅ + O ₂ ⇌ C ₂ H ₄ + HO ₂	2.000E+12	0.00	20.9	[9]	
86f	C ₂ H ₅ ⇌ C ₂ H ₄ + H	<i>k</i> ₀	3.990E+33	-4.99	167	[21]
		<i>k</i> _∞	1.110E+10	1.04	154	
87f	C ₂ H ₄ + H ⇌ C ₂ H ₃ + H ₂	4.490E+07	2.12	55.9	[22]	
88f	C ₂ H ₄ + OH ⇌ C ₂ H ₃ + H ₂ O	5.530E+05	2.31	12.4	[22]	
89f	C ₂ H ₄ + O ⇌ CH ₃ + CHO	2.250E+06	2.08	0	[12]	
90f	C ₂ H ₄ + O ⇌ CH ₂ CHO + H	1.210E+06	2.08	0	[12]	
91f	2 C ₂ H ₄ ⇌ C ₂ H ₃ + C ₂ H ₅	5.010E+14	0.00	271	[23]	
92f	C ₂ H ₄ + O ₂ ⇌ C ₂ H ₃ + HO ₂	4.220E+13	0.00	241	[24]	

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93f	$\text{C}_2\text{H}_4 + \text{HO}_2 \rightleftharpoons \text{C}_2\text{H}_4\text{O} + \text{OH}$	2.230E+12	0.00	71.9	[12]	
s93f	$\text{C}_2\text{H}_4\text{O} + \text{HO}_2 \rightleftharpoons \text{CH}_3 + \text{CO} + \text{H}_2\text{O}_2$	4.000E+12	0.00	71.2	[12]	
94f ^a	$\text{C}_2\text{H}_4 + \text{M} \rightleftharpoons \text{C}_2\text{H}_3 + \text{H} + \text{M}$	2.600E+17	0.00	404	[15]	
95f ^a	$\text{C}_2\text{H}_4 + \text{M} \rightleftharpoons \text{C}_2\text{H}_2 + \text{H}_2 + \text{M}$	3.500E+16	0.00	299	[15]	
96f	$\text{C}_2\text{H}_3 + \text{H} \rightleftharpoons \text{C}_2\text{H}_2 + \text{H}_2$	1.210E+13	0.00	0	[15]	
97f ^{a,b}	$\text{C}_2\text{H}_3 + \text{M} \rightleftharpoons \text{C}_2\text{H}_2 + \text{H} + \text{M}$	k_0	1.510E+14	0.10	137	[25]
			k_∞	6.380E+09	1.00	157
98f	$\text{C}_2\text{H}_3 + \text{O}_2 \rightleftharpoons \text{CH}_2\text{O} + \text{CHO}$	1.700E+29	-5.31	27.2	[26]	
99f	$\text{C}_2\text{H}_3 + \text{O}_2 \rightleftharpoons \text{CH}_2\text{CHO} + \text{O}$	7.000E+14	-0.61	22	[25, 26]	
100f	$\text{C}_2\text{H}_3 + \text{O}_2 \rightleftharpoons \text{C}_2\text{H}_2 + \text{HO}_2$	5.190E+15	-1.26	13.9	[25, 26]	
101f	$\text{CH}_2\text{CHO} \rightleftharpoons \text{CH}_2\text{CO} + \text{H}$	1.047E+37	-7.19	186	[24]	
102f	$\text{C}_2\text{H}_2 + \text{O} \rightleftharpoons \text{HCCO} + \text{H}$	4.000E+14	0.00	44.6	[18]	
103f	$\text{C}_2\text{H}_2 + \text{O} \rightleftharpoons \text{T-CH}_2 + \text{CO}$	1.600E+14	0.00	41.4	[18]	
104f	$\text{C}_2\text{H}_2 + \text{O}_2 \rightleftharpoons \text{CH}_2\text{O} + \text{CO}$	4.600E+15	-0.54	188	[27]	
105f	$\text{C}_2\text{H}_2 + \text{OH} \rightleftharpoons \text{CH}_2\text{CO} + \text{H}$	1.900E+07	1.70	4.18	[7, 28]	
106f	$\text{C}_2\text{H}_2 + \text{OH} \rightleftharpoons \text{C}_2\text{H} + \text{H}_2\text{O}$	3.370E+07	2.00	58.6	[7, 28]	
107f	$\text{CH}_2\text{CO} + \text{H} \rightleftharpoons \text{CH}_3 + \text{CO}$	1.110E+07	2.00	8.37	[7, 28]	
108f	$\text{CH}_2\text{CO} + \text{O} \rightleftharpoons \text{T-CH}_2 + \text{CO}_2$	2.000E+13	0.00	9.6	[7, 28]	
109f	$\text{CH}_2\text{CO} + \text{O} \rightleftharpoons \text{HCCO} + \text{OH}$	1.000E+13	0.00	8.37	[7, 28]	
110f	$\text{CH}_2\text{CO} + \text{OH} \rightleftharpoons \text{CH}_2\text{OH} + \text{CO}$	1.020E+13	0.00	0	[7, 28]	
111f	$\text{CH}_2\text{CO} + \text{CH}_3 \rightleftharpoons \text{C}_2\text{H}_5 + \text{CO}$	9.000E+10	0.00	0	[7, 28]	
112f	$\text{HCCO} + \text{H} \rightleftharpoons \text{S-CH}_2 + \text{CO}$	1.500E+14	0.00	0	[18]	
113f	$\text{HCCO} + \text{OH} \rightleftharpoons \text{CHO} + \text{CO} + \text{H}$	2.000E+12	0.00	0	[29]	
114f	$\text{HCCO} + \text{O} \rightleftharpoons 2 \text{CO} + \text{H}$	9.640E+13	0.00	0	[18]	
115f	$\text{HCCO} + \text{O}_2 \rightleftharpoons 2 \text{CO} + \text{OH}$	2.880E+07	1.70	4.19	[25]	
116f	$\text{HCCO} + \text{O}_2 \rightleftharpoons \text{CO}_2 + \text{CO} + \text{H}$	1.400E+07	1.70	4.19	[25]	
117f	$\text{C}_2\text{H} + \text{OH} \rightleftharpoons \text{HCCO} + \text{H}$	2.000E+13	0.00	0	[9, 28]	
118f	$\text{C}_2\text{H} + \text{O} \rightleftharpoons \text{CO} + \text{CH}$	1.020E+13	0.00	0	[9, 28]	
119f	$\text{C}_2\text{H} + \text{O}_2 \rightleftharpoons \text{HCCO} + \text{O}$	6.020E+11	0.00	0	[9, 28]	
120f	$\text{C}_2\text{H} + \text{O}_2 \rightleftharpoons \text{CH} + \text{CO}_2$	4.500E+15	0.00	105	[9, 28]	
121f	$\text{C}_2\text{H} + \text{O}_2 \rightleftharpoons \text{CHO} + \text{CO}$	2.410E+12	0.00	0	[9, 28]	
122f	$\text{C}_2\text{H}_2 + \text{S-CH}_2 \rightleftharpoons \text{C}_3\text{H}_3 + \text{H}$	8.000E+13	0.00	0	[17]	
123f	$\text{C}_2\text{H}_2 + \text{S-CH}_2 \rightleftharpoons \text{C}_3\text{H}_4$	8.000E+13	0.00	0	[17]	
124f	$\text{C}_2\text{H}_2 + \text{T-CH}_2 \rightleftharpoons \text{C}_3\text{H}_4$	1.200E+13	0.00	27.7	[17]	
125f	$\text{C}_2\text{H}_2 + \text{CH}_3 \rightleftharpoons \text{C}_3\text{H}_4 + \text{H}$	6.740E+19	-2.10	132	[17]	
126f	$\text{C}_3\text{H}_4 + \text{O} \rightleftharpoons \text{CH}_2\text{O} + \text{C}_2\text{H}_2$	1.000E+12	0.00	0	[29]	

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127f	$\text{C}_3\text{H}_4 + \text{O} \rightleftharpoons \text{CHO} + \text{C}_2\text{H}_3$	1.000E+12	0.00	0	[29]
128f	$\text{C}_3\text{H}_4 + \text{OH} \rightleftharpoons \text{CH}_2\text{O} + \text{C}_2\text{H}_3$	1.000E+12	0.00	0	[29]
129f	$\text{C}_3\text{H}_4 + \text{OH} \rightleftharpoons \text{CHO} + \text{C}_2\text{H}_4$	1.000E+12	0.00	0	[29]
130f	$\text{C}_3\text{H}_4 \rightleftharpoons \text{C}_3\text{H}_3 + \text{H}$	5.000E+14	0.00	370	[19]
131f	$\text{C}_3\text{H}_5 \rightleftharpoons \text{C}_3\text{H}_4 + \text{H}$	3.980E+13	0.00	293	[29]
132f	$\text{C}_3\text{H}_5 + \text{H} \rightleftharpoons \text{C}_3\text{H}_4 + \text{H}_2$	1.000E+13	0.00	0	[29]
133f	$\text{C}_3\text{H}_5 + \text{O}_2 \rightleftharpoons \text{C}_3\text{H}_4 + \text{HO}_2$	6.000E+11	0.00	41.9	[29]
134f	$\text{C}_2\text{H}_4 + \text{S-CH}_2 \rightleftharpoons \text{C}_3\text{H}_6$	6.600E+13	0.00	0	[29]
135f	$\text{C}_2\text{H}_4 + \text{T-CH}_2 \rightleftharpoons \text{C}_3\text{H}_6$	1.800E+10	0.00	0	[29]
136f	$\text{C}_3\text{H}_5 + \text{H} \rightleftharpoons \text{C}_3\text{H}_6$	k_0 1.330E+60 k_∞ 2.000E+14	-12.00 0.00	25 0	[30]
137f	$\text{C}_2\text{H}_3 + \text{CH}_3 \rightleftharpoons \text{C}_3\text{H}_6$	k_0 4.270E+58 k_∞ 2.500E+13	-11.94 0.00	40.9 0	[30]
a137f	$\text{C}_3\text{H}_6 \rightleftharpoons \text{C}_2\text{H}_2 + \text{CH}_4$	3.500E+12	0.00	293	[29]
138f	$\text{H} + \text{C}_3\text{H}_6 \rightleftharpoons \text{C}_3\text{H}_5 + \text{H}_2$	5.000E+12	0.00	6.3	[29]
139f	$\text{C}_3\text{H}_6 + \text{O} \rightleftharpoons \text{C}_2\text{H}_4 + \text{CH}_2\text{O}$	5.900E+13	0.00	21	[29]
140f	$\text{C}_3\text{H}_6 + \text{O} \rightleftharpoons \text{C}_2\text{H}_5 + \text{CHO}$	3.600E+12	0.00	0	[29]
141f	$\text{C}_3\text{H}_6 + \text{OH} \rightleftharpoons \text{C}_2\text{H}_5 + \text{CH}_2\text{O}$	7.900E+12	0.00	0	[29]
142f	$\text{C}_3\text{H}_6 + \text{OH} \rightleftharpoons \text{C}_3\text{H}_5 + \text{H}_2\text{O}$	4.000E+12	0.00	0	[29]
143f	$\text{CH}_3 + \text{C}_3\text{H}_6 \rightleftharpoons \text{CH}_4 + \text{C}_3\text{H}_5$	8.960E+12	0.00	35.6	[29]
144f	$\text{C}_3\text{H}_6 + \text{C}_2\text{H}_5 \rightleftharpoons \text{C}_3\text{H}_5 + \text{C}_2\text{H}_6$	1.000E+11	0.00	38.5	[29]
145f	$\text{N-C}_3\text{H}_7 \rightleftharpoons \text{CH}_3 + \text{C}_2\text{H}_4$	k_0 5.490E+49 k_∞ 1.230E+13	-10.00 -0.10	150 126	[31]
146f	$\text{N-C}_3\text{H}_7 \rightleftharpoons \text{H} + \text{C}_3\text{H}_6$	k_0 7.881E+39 k_∞ 1.674E+14	-6.66 0.00	178 162	[30]
147f	$\text{N-C}_3\text{H}_7 + \text{O}_2 \rightleftharpoons \text{C}_3\text{H}_6 + \text{HO}_2$	9.000E+10	0.00	0	[32, 30]
p1f	$\text{C}_3\text{H}_8 \rightleftharpoons \text{CH}_3 + \text{C}_2\text{H}_5$	k_0 7.830E+18 k_∞ 1.100E+17	0.00 0.00	272 353	[15]
p4f	$\text{C}_3\text{H}_8 + \text{O}_2 \rightleftharpoons \text{I-C}_3\text{H}_7 + \text{HO}_2$	4.000E+13	0.00	199	[33]
p5f	$\text{C}_3\text{H}_8 + \text{O}_2 \rightleftharpoons \text{N-C}_3\text{H}_7 + \text{HO}_2$	4.000E+13	0.00	199	[33]
p6f	$\text{C}_3\text{H}_8 + \text{H} \rightleftharpoons \text{I-C}_3\text{H}_7 + \text{H}_2$	1.300E+06	2.40	18.7	[33]
p7f	$\text{C}_3\text{H}_8 + \text{H} \rightleftharpoons \text{N-C}_3\text{H}_7 + \text{H}_2$	1.330E+06	2.54	28.3	[32, 31]
p8f	$\text{C}_3\text{H}_8 + \text{O} \rightleftharpoons \text{I-C}_3\text{H}_7 + \text{OH}$	4.760E+04	2.71	8.82	[32, 30]
p9f	$\text{C}_3\text{H}_8 + \text{O} \rightleftharpoons \text{N-C}_3\text{H}_7 + \text{OH}$	1.900E+05	2.68	15.6	[32, 30]
p10f	$\text{C}_3\text{H}_8 + \text{OH} \rightleftharpoons \text{I-C}_3\text{H}_7 + \text{H}_2\text{O}$	4.670E+07	1.61	-0.146	[33]
p11f	$\text{C}_3\text{H}_8 + \text{OH} \rightleftharpoons \text{N-C}_3\text{H}_7 + \text{H}_2\text{O}$	1.054E+10	0.97	6.64	[33]

Number	Reaction	A	n	E	Ref.	
p12f	$C_3H_8 + HO_2 \rightleftharpoons I-C_3H_7 + H_2O_2$	9.640E+03	2.60	58.2	[32, 31, 30]	
p13f	$C_3H_8 + HO_2 \rightleftharpoons N-C_3H_7 + H_2O_2$	4.760E+04	2.55	69	[32, 31, 30]	
p196f	$I-C_3H_7 + C_3H_8 \rightleftharpoons N-C_3H_7 + C_3H_8$	8.400E-03	4.20	36.3	[32, 34]	
p17f	$I-C_3H_7 \rightleftharpoons C_3H_6 + H$	k_0 k_∞	2.167E+17 8.760E+07	0.00 1.76	118 149	[15, 31]
p19f	$I-C_3H_7 + O_2 \rightleftharpoons C_3H_6 + HO_2$	1.300E+11	0.00	0	[32, 30]	

Units are mol, cm³, kJ, K.

The backward rates for all reversible reactions can be calculated from thermodynamic data.

^aThird-body efficiencies are:

$$[M2] = 2.5 [H_2] + 12 [H_2O] + 1.9 [CO] + 3.8 [CO_2] + 1 [\text{other}].$$

$$[M1] = 2.5 [H_2] + 16.3 [H_2O] + 1.9 [CO] + 3.8 [CO_2] + 1 [\text{other}].$$

$$[M6] = 0.5 [AR] + 0.3 [O_2] + 12 [H_2O] + 0.75 [CO] + 1.5 [CO_2] + 1.5 [C_2H_6] + 1 [\text{other}].$$

$$[M] = 1 [\text{other}].$$

$$[M7] = 0.7 [AR] + 2 [H_2] + 6 [H_2O] + 1.5 [CO] + 2 [CO_2] + 2 [CH_4] + 3 [C_2H_6] + 1 [\text{other}].$$

$$[M4] = 1.9 [H_2] + 12 [H_2O] + 2.5 [CO] + 2.5 [CO_2] + 1 [\text{other}].$$

$$[M5] = 2.4 [H_2] + 15.4 [H_2O] + 1.8 [CO] + 3.6 [CO_2] + 1 [\text{other}].$$

The third-body efficiency for H₂O in M6 is taken from [35].

^bPressure dependent reactions are described by the TROE-formulation [36]. The centering parameters are given by:

$$F_{c,8f} = \exp(-T/345 \text{ K}) + \exp(-345 \text{ K}/T).$$

$$F_{c,14f} = 0.265 \exp(-T/94 \text{ K}) + 0.735 \exp(-T/1756 \text{ K}) + \exp(-5182 \text{ K}/T).$$

$$F_{c,47f} = 0.37 \exp(-T/61 \text{ K}) + 0.63 \exp(-T/3315 \text{ K}).$$

$$F_{c,48f} = 0.38 \exp(-T/73 \text{ K}) + 0.62 \exp(-T/1180 \text{ K}).$$

$$F_{c,81f} = 0.16 \exp(-T/125 \text{ K}) + 0.84 \exp(-T/2219 \text{ K}) + \exp(-6882 \text{ K}/T).$$

$$F_{c,86f} = 0.832 \exp(-T/1203 \text{ K}).$$

$$F_{c,97f} = 0.7.$$

$$F_{c,136f} = 0.98 \exp(-T/1097 \text{ K}) + 0.02 \exp(-T/1097 \text{ K}) + \exp(-6860 \text{ K}/T).$$

$$F_{c,137f} = 0.825 \exp(-T/1341 \text{ K}) + 0.175 \exp(-T/60000 \text{ K}) + \exp(-10140 \text{ K}/T).$$

$$F_{c,145f} = 2.17 \exp(-T/251 \text{ K}) + \exp(-1185 \text{ K}/T).$$

$$F_{c,146f} = \exp(-T/1310 \text{ K}) + \exp(-48097 \text{ K}/T).$$

$$F_{c,p1f} = 0.76 \exp(-T/38 \text{ K}) + 0.24 \exp(-T/1946 \text{ K}).$$

$$F_{c,p17f} = \exp(-T/260 \text{ K}) + \exp(-3000 \text{ K}/T).$$

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